

Density-functional theory or how to connect chemistry with convex analysis

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N -electron Schrödinger equation for the *ground state*

$$\hat{H}\Psi_0 = E_0\Psi_0$$



where $\Psi_0 \equiv \Psi_0(\mathbf{r}_1, \mathbf{r}_2, \dots, \mathbf{r}_N)$ and $\hat{H} = \hat{T} + \hat{W}_{ee} + \hat{V}$ with

$$\hat{T} \equiv -\frac{1}{2} \sum_{i=1}^N \nabla_{\mathbf{r}_i}^2 = -\frac{1}{2} \sum_{i=1}^N \left(\frac{\partial^2}{\partial x_i^2} + \frac{\partial^2}{\partial y_i^2} + \frac{\partial^2}{\partial z_i^2} \right) \quad \rightarrow \quad \text{universal kinetic energy operator}$$

$$\hat{W}_{ee} \equiv \sum_{i < j}^N \frac{1}{|\mathbf{r}_i - \mathbf{r}_j|} \times \quad \rightarrow \quad \text{universal two-electron repulsion operator}$$

$$\hat{V} \equiv \sum_{i=1}^N v(\mathbf{r}_i) \times \quad \text{where} \quad v(\mathbf{r}) = - \sum_A^{\text{nuclei}} \frac{Z_A}{|\mathbf{r} - \mathbf{R}_A|} \quad \rightarrow \quad \text{local nuclear potential operator}$$

Potential-functional theory

- The basic variable in quantum mechanics is the (many-electron) *wavefunction* Ψ .
- Normalization condition:

$$1 = \langle \Psi | \Psi \rangle = \int_{\mathbb{R}^3} d\mathbf{r}_1 \dots \int_{\mathbb{R}^3} d\mathbf{r}_N \Psi^*(\mathbf{r}_1, \dots, \mathbf{r}_N) \Psi(\mathbf{r}_1, \dots, \mathbf{r}_N)$$

- Rayleigh–Ritz *variational principle*:

$$E_0 = \min_{\Psi} \langle \Psi | \hat{H} | \Psi \rangle = \langle \Psi_0 | \hat{H} | \Psi_0 \rangle$$

where

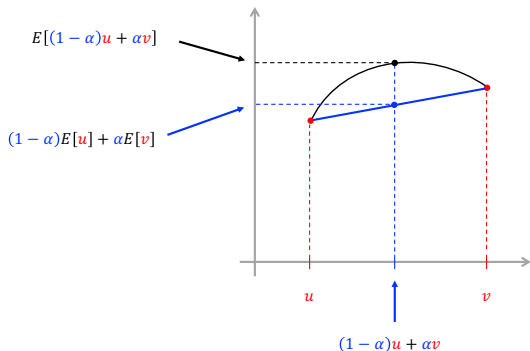
$$\langle \Psi | \hat{H} | \Psi \rangle = \int_{\mathbb{R}^3} d\mathbf{r}_1 \dots \int_{\mathbb{R}^3} d\mathbf{r}_N \Psi^*(\mathbf{r}_1, \dots, \mathbf{r}_N) \hat{H} \Psi(\mathbf{r}_1, \dots, \mathbf{r}_N)$$

- The ground-state energy can be seen as a *functional* of the nuclear potential $v : \mathbf{r} \mapsto v(\mathbf{r})$,

$$v \quad \longrightarrow \quad \hat{H} \equiv \hat{H}[v] \quad \longrightarrow \quad \Psi_0 \equiv \Psi[v] \quad \longrightarrow \quad E_0 \equiv E[v]$$

Potential-functional theory

- The ground-state energy is a *concave* functional of the nuclear potential.



Proof: since $\hat{H}[(1-\alpha)u + \alpha v] = (1-\alpha)\hat{H}[u] + \alpha\hat{H}[v]$ we have

$$\begin{aligned} E[(1-\alpha)u + \alpha v] &= (1-\alpha) \langle \Psi[(1-\alpha)u + \alpha v] | \hat{H}[u] | \Psi[(1-\alpha)u + \alpha v] \rangle \\ &\quad + \alpha \langle \Psi[(1-\alpha)u + \alpha v] | \hat{H}[v] | \Psi[(1-\alpha)u + \alpha v] \rangle \\ &\geq (1-\alpha)E[u] + \alpha E[v] \end{aligned}$$

Potential-functional theory

- We consider two potentials u and v that *differ by more than a constant*.
- In other words $u(\mathbf{r}) - v(\mathbf{r})$ varies with \mathbf{r} .
- The corresponding Hamiltonians $\hat{H}[u]$ and $\hat{H}[v]$ have no eigenfunctions in common.

Proof: if Ψ is eigenfunction of both $\hat{H}[u]$ and $\hat{H}[v]$ then

$$\begin{aligned}\left(\hat{H}[u] - \hat{H}[v]\right)\Psi(\mathbf{r}, \mathbf{r}, \dots, \mathbf{r}) &= N(u(\mathbf{r}) - v(\mathbf{r})) \times \Psi(\mathbf{r}, \mathbf{r}, \dots, \mathbf{r}) \\ &= (E[u] - E[v]) \times \Psi(\mathbf{r}, \mathbf{r}, \dots, \mathbf{r})\end{aligned}$$

thus leading to $u(\mathbf{r}) - v(\mathbf{r}) = \frac{E[u] - E[v]}{N}$, which is absurd.

Density-functional theory (DFT)

- Electron density:

$$n_v(\mathbf{r}) = N \int_{\mathbb{R}^3} d\mathbf{r}_2 \dots \int_{\mathbb{R}^3} d\mathbf{r}_N \left| \Psi[v](\mathbf{r}, \mathbf{r}_2, \dots, \mathbf{r}_N) \right|^2$$

- Note that n_v is a so-called “*v*-representable” density.
- A density n is said to be “*N*-representable” if you can find an N -electron wavefunction Ψ such that

$$n(\mathbf{r}) = N \int_{\mathbb{R}^3} d\mathbf{r}_2 \dots \int_{\mathbb{R}^3} d\mathbf{r}_N \left| \Psi(\mathbf{r}, \mathbf{r}_2, \dots, \mathbf{r}_N) \right|^2$$

- It can be shown that the set of N -representable densities is the set of positive functions $n : \mathbf{r} \mapsto n(\mathbf{r})$ that integrate to the number of electrons N and that give a **finite von Weizsäcker kinetic energy***:

$$\frac{1}{2} \int d\mathbf{r} \left| \nabla n^{1/2}(\mathbf{r}) \right|^2 < +\infty.$$

* *Principles of DFT*, lecture given by **Trygve Helgaker** at the GDR Correl mini-school on mathematics in electronic structure theory, Paris, January 2017, http://folk.uio.no/helgaker/talks/Paris_2017.pdf

Density-functional theory

- The *nuclear potential energy* (unlike the kinetic and repulsion energies) is an *explicit* functional of the density.

Proof: Since electrons are indistinguishable particles,

$$\left| \Psi(\mathbf{r}_1, \dots, \mathbf{r}_i, \dots, \mathbf{r}_j, \dots, \mathbf{r}_N) \right|^2 = \left| \Psi(\mathbf{r}_1, \dots, \mathbf{r}_j, \dots, \mathbf{r}_i, \dots, \mathbf{r}_N) \right|^2,$$

thus leading to

$$\begin{aligned} \langle \Psi | \hat{V} | \Psi \rangle &= \sum_{i=1}^N \int_{\mathbb{R}^3} d\mathbf{r}_1 \dots \int_{\mathbb{R}^3} d\mathbf{r}_N v(\mathbf{r}_i) \left| \Psi(\mathbf{r}_1, \dots, \mathbf{r}_i, \dots, \mathbf{r}_N) \right|^2 \\ &= N \int_{\mathbb{R}^3} d\mathbf{r}_1 \dots \int_{\mathbb{R}^3} d\mathbf{r}_N v(\mathbf{r}_1) \left| \Psi(\mathbf{r}_1, \dots, \mathbf{r}_N) \right|^2 \\ &= \int_{\mathbb{R}^3} d\mathbf{r}_1 v(\mathbf{r}_1) n(\mathbf{r}_1) \\ \underline{\underline{\text{notation}}} & \quad (v|n) \end{aligned}$$

Hohenberg–Kohn theorem

- Let us consider two potentials u and v that *differ by more than a constant*.
- According to the variational principle, the following (strict) inequalities hold:

$$E[u] < \langle \Psi[v] | \hat{H}[u] | \Psi[v] \rangle \quad \text{and} \quad E[v] < \langle \Psi[u] | \hat{H}[v] | \Psi[u] \rangle,$$

or, equivalently,

$$E[u] < E[v] + (u - v|n_v) \quad \text{and} \quad E[v] < E[u] - (u - v|n_u),$$

thus leading to

$$(u - v|n_u - n_v) < 0$$

- **Important consequence** of the above strict inequality:

$$n_u \neq n_v \quad \longleftarrow \text{Hohenberg–Kohn theorem}^*$$

*P. Hohenberg and W. Kohn, Phys. Rev. 136, B864 (1964).

Density-functional theory

- The theorem is usually formulated as follows:

“There is a one-to-one correspondence between the local potential and the ground-state density”

- Let us return to the following inequality:

$$E[v] < E[u] - (u - v|n_u).$$

- If we now include the (missing) particular case where u and v differ by a constant, then

$$E[v] \leq E[u] - (u - v|n_u).$$

- Note that the latter inequality can be seen as holding for *“any”* v and for a *fixed* u .
- **Universal** Hohenberg–Kohn density functional and Lieb maximization:*

$$E[u] - (u|n_u) \equiv F[n_u] = \max_v \left\{ E[v] - (v|n_u) \right\}$$

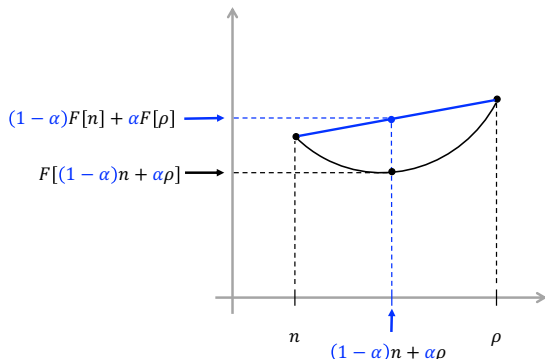
* E. Lieb, Int. J. Quantum Chem. **24**, 243 (1983).

Hohenberg–Kohn variational principle and convexity

- In DFT, if you know $F[n]$, you know “everything”:

$$F[n] = \max_v \{E[v] - (v|n)\} \Leftrightarrow E[v] = \min_n \{F[n] + (v|n)\}$$

- Note that the *density* has become our *basic variable*.
- Note also that $F[n]$ is a *convex* functional of the density.



Proof:

$$\begin{aligned} E[v] - (v|(1-\alpha)n + \alpha\rho) &= (1-\alpha)(E[v] - (v|n)) + \alpha(E[v] - (v|\rho)) \\ &\leq (1-\alpha)F[n] + \alpha F[\rho] \end{aligned}$$

- Let us now consider DFT for *non-interacting* electrons:

$$\hat{W}_{ee} \longrightarrow 0$$

$$F[n] \longrightarrow T_s[n] \leftarrow \text{non-interacting kinetic energy density functional}$$

- In a practical (so-called Kohn–Sham) DFT calculation the following (so-called *exchange–correlation*) energy contribution is described as a (approximate) density functional:

$$E_{xc}[n] = F[n] - T_s[n] - \frac{1}{2} \int_{\mathbb{R}^3} \int_{\mathbb{R}^3} d\mathbf{r} d\mathbf{r}' \frac{n(\mathbf{r})n(\mathbf{r}')}{|\mathbf{r} - \mathbf{r}'|}.$$

- In Kohn–Sham (KS) DFT, the many-electron problem is formally transformed into a *one-electron-like* problem.
- The “magical” one-electron wavefunctions (so-called KS orbitals) fulfill the following *self-consistent* KS equations:

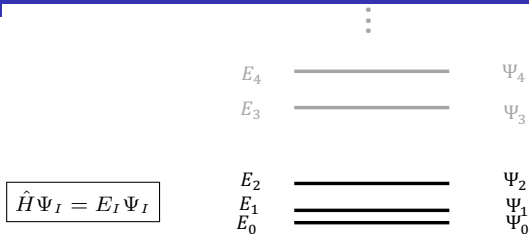
$$\left(-\frac{1}{2}\nabla_{\mathbf{r}}^2 + v(\mathbf{r}) + \int_{\mathbb{R}^3} d\mathbf{r}' \frac{n(\mathbf{r}')}{|\mathbf{r} - \mathbf{r}'|} + \frac{\delta E_{\text{xc}}[n]}{\delta n(\mathbf{r})} \right) \varphi_i(\mathbf{r}) = \varepsilon_i \varphi_i(\mathbf{r})$$

where $n(\mathbf{r}) = \sum_{i=1}^N |\varphi_i(\mathbf{r})|^2$.

- When convergence is reached, the (in-principle-exact) ground-state energy reads

$$E[v] = -\frac{1}{2} \sum_{i=1}^N \int_{\mathbb{R}^3} d\mathbf{r} \varphi_i^*(\mathbf{r}) \nabla_{\mathbf{r}}^2 \varphi_i(\mathbf{r}) + (v|n) + \frac{1}{2} \int_{\mathbb{R}^3} \int_{\mathbb{R}^3} d\mathbf{r} d\mathbf{r}' \frac{n(\mathbf{r})n(\mathbf{r}')}{|\mathbf{r} - \mathbf{r}'|} + E_{\text{xc}}[n].$$

DFT for an ensemble of states



- An ensemble is characterized by the *fixed weights* that are assigned to the excited states:

$$w_1 \geq w_2 \geq \dots \geq w_M \geq 0.$$

- In this context the *energy of the ensemble*

$$E^{\{w_1, \dots, w_M\}} = E_0 + \sum_{I=1}^M w_I (E_I - E_0)$$

becomes a *functional of the ensemble density**

$$n^{\{w_1, \dots, w_M\}}(\mathbf{r}) = n_0(\mathbf{r}) + \sum_{I=1}^M w_I (n_I(\mathbf{r}) - n_0(\mathbf{r})).$$

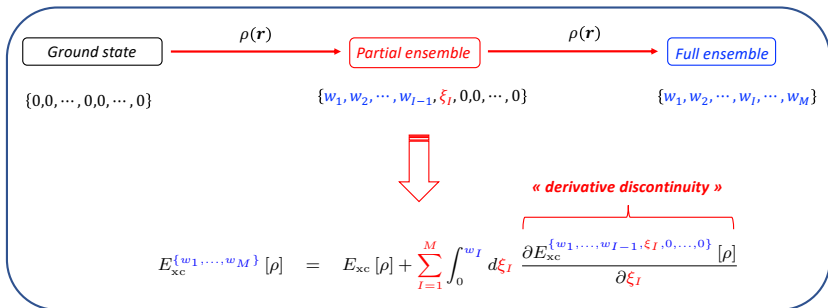
*E. K. U. Gross, L. N. Oliveira, and W. Kohn, Phys. Rev. A 37, 2809 (1988).

DFT for an ensemble of states

- In ensemble DFT, the exchange-correlation functional becomes *weight-dependent*:

$$E_{\text{xc}}[n] = E_{\text{xc}}^{\{w_1=0, \dots, w_M=0\}}[n] \longrightarrow E_{\text{xc}}^{\{w_1, \dots, w_M\}}[n]$$

- This weight dependence plays a crucial role in the calculation of *excited energy levels*.¹
- Generalized adiabatic connection formalism for ensembles (GACE):^{2,3}



¹K. Deur and E. Fromager, J. Chem. Phys. **150**, 094106 (2019).

²O. Franck and E. Fromager, Mol. Phys. **112**, 1684 (2014).

³K. Deur, L. Mazouin, and E. Fromager, Phys. Rev. B **95**, 035120 (2017).

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